

# 7th Grade Science Agenda- Mrs. Sharon

Week of December 5, 2016

Day	In Class/Learning Targets	HW/Reminders
<p><b>Monday 12-5</b></p> <p><i>I can describe physical and chemical properties of matter.</i></p>	<p><b>See Math and Science</b></p> <p><b>Check: Candy Compounds Lab</b></p> <ol style="list-style-type: none"> <li>Physical vs. Chemical Properties -class discussion/notes in notebook -evidence of chemical reactions</li> <li>Egg in Vinegar Demo</li> <li>Rapid Rusting Lab</li> </ol>	<p><b>Success Criteria</b></p> <p>-8/10 physical/chemical properties</p> <p>-Identifying evidence of chemical properties</p>
<p><b>Tuesday 12-6</b></p>	<p><b>Block Schedule-Odd Day (2, 4, 6)</b></p>	
<p><b>Wednesday 12-7</b></p> <p><i>I can observe physical and chemical properties before and after a chemical reaction.</i></p>	<p><b>Block Schedule-Odd Day (1, 3, 7)</b></p> <ol style="list-style-type: none"> <li>Mystery Powders Lab</li> </ol>	<p><b>Success Criteria</b></p> <p>-Identifying the mystery powder based on chemical and physical properties</p>
<p><b>Thursday 12-8</b></p>	<p><b>Block Schedule-Even Day (2, 4, 6)</b></p> <p><b>See Wednesday</b></p>	
<p><b>Friday 12-9</b></p> <p><i>I can observe physical and chemical properties before and after a chemical reaction.</i></p>	<p><b>See All Classes-Early Release</b></p> <ol style="list-style-type: none"> <li>Observe Egg in Vinegar</li> <li>Finish Rapid Rusting Lab</li> <li>Finish Mystery Powders Lab</li> </ol>	<p><b>Success Criteria:</b></p> <p>-Identifying properties before and after the chemical reactions</p> <p>-Identifying the mystery powders</p>

Standards Covered This Week:

**MS-PS1-1 Develop models to describe the atomic composition of simple molecules and extended structures.**

**PS1.A: Disciplinary Core Ideas**

- Substances are made from different types of atoms, which combine with one another in various ways. Atoms form molecules that range in size from two to thousands of atoms.
- Gases and liquids are made of molecules or inert atoms that are moving relative to each other.
- In a liquid, the molecules are constantly in contact with others; in a gas, they are widely spaced except when they happen to collide. In a solid, atoms are closely spaced and may vibrate in position but do not change relative locations
- Solids may be formed from molecules, or they may be extended structures with repeating subunits (e.g., crystals).
- The changes of state that occur with variations in temperature or pressure can be described and predicted using these models of matter.

**MS-PS1-2 Analyze and interpret data on the properties of substances before and after the substances interact to determine if a chemical reaction has occurred.**

**PS1.A: Disciplinary Core Ideas**

**Structure and Properties of Matter**

- Each pure substance has characteristic physical and chemical properties (for any bulk quantity under given conditions) that can be used to identify it.

**Patterns**

- Macroscopic patterns are related to the nature of microscopic and atomic-level structure.