

7th Grade Science Agenda- Mrs. Sharon

Week of December 19, 2016

Day	In Class/Learning Targets	HW/Reminders
<p>Monday 12-19</p> <p><i>I can identify physical and chemical properties of matter before and after chemical reactions.</i></p>	<p>Block Schedule-Odd Day (1, 3, 7)</p> <p>Turn In: Study Guide</p> <ol style="list-style-type: none"> 1. Chemistry Unit Test Holiday Crystal Lab #15-18 Pass Back Papers Check Mistar and Clean out Binders 	
<p>Tuesday 12-20</p>	<p>Block Schedule-Even Day (2, 4, 6)</p> <p>See Monday</p>	
<p>Wednesday 12-21</p> <p><i>I can identify physical and chemical properties of matter before and after chemical reactions.</i></p>	<p>See All Classes</p> <ol style="list-style-type: none"> 1. Finish Holiday Crystals Lab #19-23 2. Wrap crystals and take home 3. Periodic Table Holiday Picture 	
<p>Thursday 12-22</p>	<p>WINTER BREAK No School</p>	
<p>Friday 12-23</p>	<p>WINTER BREAK No School</p> <p>Have a fun, safe and relaxing break with your families! Merry Christmas and Happy New Year!</p>	<p>See you next year! Classes resume January 4, 2017.</p>

Standards Covered This Unit:

MS-PS1-1 Develop models to describe the atomic composition of simple molecules and extended structures.

PS1.A: Disciplinary Core Ideas

- Substances are made from different types of atoms, which combine with one another in various ways. Atoms form molecules that range in size from two to thousands of atoms.
- Gases and liquids are made of molecules or inert atoms that are moving relative to each other.
- In a liquid, the molecules are constantly in contact with others; in a gas, they are widely spaced except when they happen to collide. In a solid, atoms are closely spaced and may vibrate in position but do not change relative locations
- Solids may be formed from molecules, or they may be extended structures with repeating subunits (e.g., crystals).
- The changes of state that occur with variations in temperature or pressure can be described and predicted using these models of matter.

MS-PS1-2 Analyze and interpret data on the properties of substances before and after the substances interact to determine if a chemical reaction has occurred.

PS1.A: Disciplinary Core Ideas

Structure and Properties of Matter

- Each pure substance has characteristic physical and chemical properties (for any bulk quantity under given conditions) that can be used to identify it.

Patterns

- Macroscopic patterns are related to the nature of microscopic and atomic-level structure.