

# 7th Grade Science Agenda- Mrs. Sharon

## Week of November 21, 2016

Day	In Class/Learning Targets	HW/Reminders
<b>Monday</b> <b>11-21</b>  <i>I can understand how the Periodic Table is organized.</i>	<b>Block Schedule (1, 3, 7)</b>  <b>Check: Metals vs. NonMetals</b>  <b>1. Element Research-Mini Poster</b> Poster Due: November 28/29  Helpful websites: <a href="http://elements.wlonk.com/ElementsTable.htm">http://elements.wlonk.com/ElementsTable.htm</a> <a href="http://www.ducksters.com/science/elements.php">http://www.ducksters.com/science/elements.php</a>	You need your laptop!  <u><b>Success Criteria</b></u> Accurately completing element research and mini poster
<b>Tuesday</b> <b>11-22</b>	<b>Block Schedule (2, 4, 6)</b>  <b>See Monday</b>	
<b>Wednesday</b> <b>11-23</b>	<b>No School</b>	
<b>Thursday</b> <b>11-24</b>	<b>No School</b> <b>Happy Thanksgiving!</b>	
<b>Friday</b> <b>11-25</b>	<b>No School</b>	

Standards Covered This Week:

**MS-PS1-1 Develop models to describe the atomic composition of simple molecules and extended structures.**

**PS1.A: Disciplinary Core Ideas**

- Substances are made from different types of atoms, which combine with one another in various ways. Atoms form molecules that range in size from two to thousands of atoms.

- Gases and liquids are made of molecules or inert atoms that are moving relative to each other.
- In a liquid, the molecules are constantly in contact with others; in a gas, they are widely spaced except when they happen to collide. In a solid, atoms are closely spaced and may vibrate in position but do not change relative locations
- Solids may be formed from molecules, or they may be extended structures with repeating subunits (e.g., crystals).
- The changes of state that occur with variations in temperature or pressure can be described and predicted using these models of matter.

**MS-PS1- 2. Analyze and interpret data on the properties of substances before and after the substances interact to determine if a chemical reaction has occurred.**

**PS1.A: Disciplinary Core Ideas  
Structure and Properties of Matter**

- Each pure substance has characteristic physical and chemical properties (for any bulk quantity under given conditions) that can be used to identify it.

**Patterns**

- Macroscopic patterns are related to the nature of microscopic and atomic-level structure.